



Efficient synthesis of *N,N'*-dialkyl-*N''*-dialkylaminocarbothioyl thioureas from cyclic secondary amines, CS₂, and *N,N'*-dialkyl carbodiimides in water

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ARTICLE INFO

Article history:

Received 12 February 2008

Revised 23 April 2008

Accepted 29 April 2008

Available online 2 May 2008

Keywords:

Dithiobiuret

Carbodiimide

Secondary amine

Aqueous media

Carbon disulfide

ABSTRACT

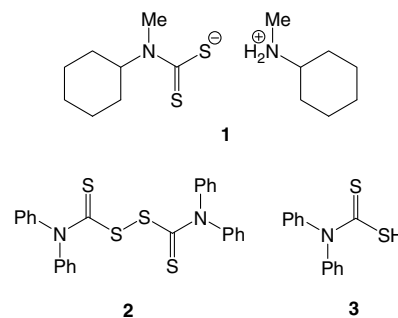
A mild, convenient, and practical one-pot procedure for direct synthesis of *N,N'*-dialkyl-*N''*-dialkylaminocarbothioyl thioureas is described via three-component reaction of cyclic secondary amines, CS₂, and *N,N'*-dialkyl carbodiimides in water at room temperature.

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Acylureas and biurets are acyclic compounds with anticonvulsant activity. Replacement of one or both oxygen atoms of these compounds with sulfur results in the formation of thiobiurets and dithiobiurets. This modification increases the lipophilicity and possibly the biological activity of these compounds. The structural similarity of dithiobiuret to biurets has allowed researchers to test several analogues for their antifungal and insecticide activities.¹ Some dithiobiurets were synthesized as male insect chemosterilants.² Also these compounds are useful reagents or intermediates in the synthesis of heterocycles and polymers with biological properties.³ Dithiobiurets are (*S,S*)-bidentate ligands and are used widely in coordination chemistry.^{4,5} Several approaches for the synthesis of dithiobiurets have been described.^{1–7}

It has been shown that the reaction of primary amines with CS₂ in Et₂O or THF leads to thioureas and/or isothiocyanates.⁸ Also, using *N*-methylcyclohexylamine or diphenylamine as secondary amines, in Et₂O or pyridine, results in salt **1** or compound **2**, respectively. Compound **3** is readily oxidized to the disulfide **2** in the presence of oxygen⁸ (Scheme 1).

Herein we report a three-component reaction between cyclic secondary amines **4** and CS₂ in the presence of *N,N'*-dialkyl carbodiimides **5** in water,^{9,10} which leads to *N,N'*-dialkyl-*N''*-dialkylaminocarbothioyl thioureas **6** in excellent yields (Scheme 2). When these reactions were carried out in the presence of acyclic secondary amines (dibenzylamine, diethylamine, dimethylamine)



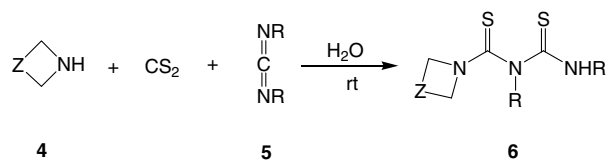
Scheme 1.

and primary amines (benzylamine, *tert*-butylamine), a complex reaction mixture was obtained.

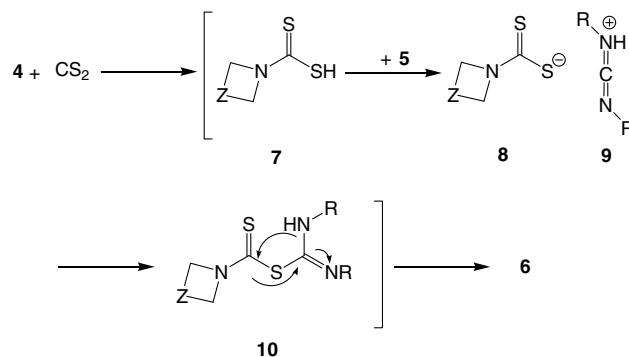
The reaction of **4**, CS₂, and **5** proceeded smoothly in H₂O, and was complete within 5 h. The ¹H NMR and ¹³C NMR spectra of the reaction mixtures after work-up clearly indicated the formation of *N,N'*-dialkyl-*N''*-dialkylaminocarbothioyl thioureas **6a–6h** in 87–96% yields (Table 1).

The structures of compounds **6a–6h** were apparent from their mass spectra, which displayed, in each case, the molecular ion peak at the appropriate *m/z* value. The ¹H and ¹³C NMR spectroscopic data, as well as the IR spectra, were in agreement with the proposed structures. For example, the ¹H NMR spectrum of **6a** showed characteristic multiplets for the methylene ($\delta = 1.05–2.00$ and 3.83) and methine ($\delta = 4.27–4.34$ and 4.93–4.99) protons, together with a doublet ($\delta = 6.03$, ³*J* = 7.5) for the NH group. The ¹³C NMR

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Scheme 2.



Scheme 3.

Table 1
Reaction of cyclic secondary amines **4**, CS₂, and *N,N'*-dialkyl carbodiimides **5** in H₂O

Entry	4	R in 5	6	Yield ^a (%)
a		Cyclohexyl		91
b		<i>i</i> -Pr		89
c		Cyclohexyl		92
d		<i>i</i> -Pr		95
e		Cyclohexyl		87
f		<i>i</i> -Pr		93
g		Cyclohexyl		95
h		Cyclohexyl		96

^a Isolated yield.

spectrum of **6a** showed 11 distinct resonances ($\delta = 24.2$ – 60.9) in the aliphatic region of the spectrum, together with two signals at $\delta = 179.0$ and 183.4 ppm for the C=S groups.

A tentative mechanism for this transformation is proposed in Scheme 3. It is conceivable that the initial event is the formation of the SH-acidic intermediate **7** from **4** and CS₂ that can protonate **5** to afford **9**. Subsequent nucleophilic attack of the sulfur anion **8** on **9** results in **10**, which undergoes a well-documented rearrangement^{11,12} to produce **6**.¹³

In conclusion, we have described a novel system that is effective for the synthesis of *N,N'*-dialkyl-*N''*-dialkylaminocarbothioyl thioureas in high yields at room temperature in water. The advantage

of the present procedure over existing methods is that the reaction is performed in aqueous media and under solvent-free conditions by simple mixing of the starting materials.

References and notes

- Pandeya, S. N.; Kumar, A.; Singh, B. N.; Mishra, D. N. *Pharm. Res.* **1987**, *4*, 321.
- Oliver, J. E.; Chang, S. C.; Brown, R. T.; Borkovec, B. J. *Med. Chem.* **1971**, *14*, 772.
- Kenawy, E.; Worley, S. D.; Broughton, R. *Biomacromolecules* **2007**, *8*, 1359.
- Crane, J. D.; Herod, A. *Inorg. Chem. Commun.* **2004**, *7*, 38.
- Billson, T. S.; Crane, J. D.; Sinn, E.; Teat, S. J.; Wheeler, E.; Young, N. A. *Inorg. Chem. Commun.* **1999**, *2*, 527.
- McGrady, J. E.; Mingos, D. M. P. *J. Chem. Soc., Perkin Trans. 2* **1996**, 355.
- Shibuya, I.; Nakanishi, H. *Bull. Chem. Soc. Jpn.* **1987**, 1381.
- Jochims, J. C. *Chem. Ber.* **1968**, *101*, 1746.
- Chan, T. H. *Organic Reactions in Aqueous Media*; Wiley: New York, 1997.
- Jun-Li, C. *Chem. Rev.* **2005**, *105*, 3095.
- Williams, A.; Ibrahim, I. T. *Chem. Rev.* **1981**, *81*, 589.
- Kurzer, F.; Douraghi-Zadeh, K. *Chem. Rev.* **1967**, *67*, 107.
- Typical procedure for the synthesis of **6**: Amine (2 mmol) was added slowly to a mixture of CS₂ (2 mmol) and *N,N'*-dialkyl carbodiimide (2 mmol) in 5 ml of water at rt. The reaction mixture was stirred for 5 h. After completion of the reaction, the resulting solid was filtered off, dried and analyzed by ¹H NMR, and ¹³C NMR. In some cases, further purification was carried out by recrystallization from CH₃CN. Compound **6a**: White powder, mp 130–134 °C, 0.33 g, yield 91%. IR (KBr) ($\nu_{\text{max}}/\text{cm}^{-1}$): 3330, 2905, 1511, 1473, 1433, 1384, 1350, 1314, 1228, 1131. MS (EI, 70 eV): *m/z* (%) = 367 (M⁺, 6), 128 (60), 98 (100), 84 (46), 55 (78), 41 (62). Anal. Calcd for C₁₉H₃₃N₃S₂ (367.61): C, 62.08; H, 9.05; N, 11.43. Found: C, 61.97; H, 8.90; N, 11.25. ¹H NMR (500.1 MHz, CDCl₃): δ 1.05–1.22 (m, 4H, 2CH₂), 1.30–1.43 (m, 4H, 2CH₂), 1.56–1.77 (m, 14H, 7CH₂), 1.92–2.00 (m, 4H, 2CH₂), 3.83 (br, 4H, 2CH₂), 4.27–4.34 (m, 1H, CH), 4.93–4.99 (m, 1H, CH), 6.03 (d, ³*J* = 7.5, 1H, NH). ¹³C NMR (125.7 MHz, CDCl₃): δ 24.2 (CH₂), 25.0 (2CH₂), 25.8 (CH₂), 25.9 (CH₂), 26.1 (2CH₂), 26.3 (2CH₂), 30.7 (2CH₂), 33.1 (2CH₂), 51.6 (2CH₂), 53.7 (CH), 60.9 (CH), 179.0 (C=S), 183.4 (C=S). Compound **6b**: Yellow powder, mp 166–170 °C, 0.25 g, yield 89%. IR (KBr) ($\nu_{\text{max}}/\text{cm}^{-1}$): 3320, 2940, 1612, 1561, 1514, 1242. Anal. Calcd for C₁₃H₂₅N₃S₂ (287.48): C, 54.31; H, 8.78; N, 14.62. Found: C, 54.11; H, 8.52; N, 14.52. ¹H NMR (500.1 MHz, CDCl₃): δ 1.20 (d, ³*J* = 6.5, 6H, 2CH₃), 1.34 (d, ³*J* = 6.6, 6H, 2CH₃), 1.68 (br, 6H, 3CH₂), 4.08 (br, 4H, 2CH₂), 4.55–4.62 (m, 1H, CH), 5.35–5.40 (m, 1H, CH), 5.87 (d, ³*J* = 6.5, 1H, NH). ¹³C NMR (125.7 MHz, CDCl₃): δ 20.2 (CH₂), 22.4 (2CH₃), 23.8 (2CH₃), 25.6 (2CH₂), 42.1 (2CH₂), 46.8 (CH), 52.7 (CH), 179.0 (C=S), 182.7 (C=S). Compound **6c**: White powder, mp 152–155 °C, 0.32 g, yield 92%. IR (KBr) ($\nu_{\text{max}}/\text{cm}^{-1}$): 3330, 2905, 1514, 1475, 1435, 1314. Anal. Calcd for C₁₈H₃₁N₃S₂ (353.58): C, 61.14; H, 8.84; N, 11.88. Found: C, 61.09; H, 8.80; N, 11.79. ¹H NMR (500.1 MHz, CDCl₃): δ 1.04–1.25 (m, 4H, 2CH₂), 1.30–1.43 (m, 4H, 2CH₂), 1.56–1.62 (m, 4H, 2CH₂), 1.72–1.82 (m, 4H, 2CH₂), 1.88–1.90 (m, 2H, CH₂), 1.97 (br, 6H, 3CH₂), 3.46–3.56 (br, 2H, CH₂), 3.57–3.87 (br, 2H, CH₂), 4.31–4.37 (m, 1H, CH), 4.90–4.95 (m, 1H, CH), 5.85 (d, ³*J* = 6.9, 1H, NH). ¹³C NMR (125.7 MHz, CDCl₃): δ 24.5 (2CH₂), 24.9 (2CH₂), 25.4 (CH₂), 25.5 (CH₂), 25.9 (3CH₂), 30.4 (2CH₂), 32.7 (2CH₂), 33.9 (CH₂), 53.0 (CH), 60.3 (CH), 178.4 (C=S), 181.6 (C=S). Compound **6d**: Yellow powder, mp 132–136 °C, 0.26 g, yield 95%. IR (KBr) ($\nu_{\text{max}}/\text{cm}^{-1}$): 3305, 2940, 1516, 1439, 1384, 1311, 1283, 1248, 1210, 1162, 1122, 1077. Anal. Calcd for C₁₂H₂₃N₃S₂ (273.45): C, 52.71; H, 8.48; N, 15.37. Found: C, 52.67; H, 8.39; N, 15.30. ¹H NMR (500.1 MHz, CDCl₃): δ 1.17 (d, ³*J* = 6.5, 6H, 2CH₃), 1.33 (d, ³*J* = 6.7, 6H, 2CH₃), 1.98 (br, 4H, 2CH₂), 3.44–3.93 (br d, 4H, 2CH₂), 4.56–4.65 (m, 1H, CH), 5.26–5.34 (m, 1H, CH), 5.68 (d, ³*J* = 6.5, 1H, NH). ¹³C NMR (125.7 MHz, CDCl₃): δ 20.6 (2CH₃), 22.5 (2CH₃), 22.6 (2CH₂), 46.6 (CH), 52.4 (2CH₂), 52.5 (CH), 178.6 (C=S), 181.0 (C=S). Compound **6e**: White powder, mp 147–150 °C, 0.32 g, yield 87%. IR (KBr) ($\nu_{\text{max}}/\text{cm}^{-1}$): 3325, 2900, 1508, 1474, 1427, 1388, 1350, 1312, 1266, 1226, 1110, 1043. Anal. Calcd for C₁₈H₃₁N₃O₂S₂ (369.58): C, 58.50; H, 8.45; N, 11.37. Found: C, 58.41; H, 8.39; N, 11.28. ¹H NMR (500.1 MHz, CDCl₃): δ 1.05–

1.21 (m, 4H, 2CH₂), 1.29–1.42 (m, 4H, 2CH₂), 1.56–1.69 (m, 4H, 2CH₂), 1.69–1.77 (m, 4H, 2CH₂), 1.87–1.90 (br, 2H, CH₂), 1.97–2.00 (br, 2H, CH₂), 3.65–3.75 (m, 4H, 2CH₂), 3.88 (br, 4H, 2CH₂), 4.26–4.33 (m, 1H, CH), 4.91–4.98 (m, 1H, CH), 5.92 (d, ³J = 7.5, 1H, NH). ¹³C NMR (125.7 MHz, CDCl₃): δ 24.5 (2CH₂), 25.3 (CH₂), 25.4 (CH₂), 25.8 (2CH₂), 30.2 (2CH₂), 32.7 (2CH₂), 50.1 (2CH₂), 53.2 (CH), 60.5 (CH), 65.9 (2CH₂), 178.9 (C=S), 184.2 (C=S). Compound **6f**: White powder, mp 123–127 °C, 0.27 g, yield 93%. IR (KBr) (ν_{max}/cm⁻¹): 3315, 2945, 2830, 1514, 1456, 1427, 1383, 1313, 1263, 1220, 1117. Anal. Calcd for C₁₂H₂₃N₃O₅S₂ (289.45): C, 49.79; H, 8.01; N, 14.52. Found: C, 49.70; H, 7.96; N, 14.49. ¹H NMR (500.1 MHz, CDCl₃): δ 1.15 (d, ³J = 6.5, 6H, 2CH₃), 1.29 (d, ³J = 6.7, 6H, 2CH₃), 3.67 (t, ³J = 4.9, 4H, 2CH₂), 3.84 (br, 4H, 2CH₂), 4.51–4.55 (m, 1H, CH), 5.27–5.33 (m, 1H, CH), 5.72 (d, ³J = 7.0, 1H, NH). ¹³C NMR (125.7 MHz, CDCl₃): δ 20.2 (2CH₃), 22.5 (2CH₃), 46.9 (CH), 50.0 (2CH₂), 52.7 (CH), 65.9 (2CH₂), 179.2 (C=S), 183.8 (C=S). Compound **6g**: Pale yellow powder, mp 59–61 °C, 0.36 g, yield 95%. IR (KBr) (ν_{max}/cm⁻¹): 3335, 2905, 1504, 1477, 1437, 1377, 1338, 1258, 1216, 727. Anal. Calcd for C₁₉H₃₃N₃O₅S₂ (383.61): C, 59.49; H, 8.67; N, 10.95.

Found: C, 59.21; H, 8.49; N, 10.79. ¹H NMR (500.1 MHz, CDCl₃): δ 1.01–1.14 (m, 4H, 2CH₂), 1.25–1.32 (m, 4H, 2CH₂), 1.50–1.71 (m, 10H, 5CH₂), 1.82–1.89 (m, 6H, 3CH₂), 2.30 (br, 1H, OH), 3.78 (br, 2H, 2CH), 3.98 (br, 3H, 3CH), 4.20 (br, 1H, CH), 4.85 (br, 1H, CH), 5.95 (d, ³J = 7.1, 1H, NH). ¹³C NMR (125.7 MHz, CDCl₃): δ 24.5 (2CH₂), 25.2 (CH₂), 25.4 (CH₂), 25.9 (3CH₂), 30.2 (CH₂), 32.6 (2CH₂), 32.3 (2CH₂), 46.6 (2CH₂), 53.3 (CH), 60.6 (CH), 64.7 (CH), 178.5 (C=S), 183.1 (C=S). Compound **6h**: Pale brown powder, mp 115–120 °C, 0.37 g, yield 96%. IR (KBr) (ν_{max}/cm⁻¹): 3340, 2905, 1521, 1472, 1328, 1270, 1214, 1128. Anal. Calcd for C₁₉H₃₃N₃O₅S₂ (383.61): C, 59.49; H, 8.67; N, 10.95. Found: C, 59.20; H, 8.51; N, 10.80. ¹H NMR (500.1 MHz, CDCl₃): δ 1.03–1.14 (m, 6H, 3CH₂), 1.15–1.22 (m, 2H, CH₂), 1.23–1.36 (m, 6H, 3CH₂), 1.62–1.81 (m, 4H, 2CH₂), 1.83–1.93 (m, 2H, CH₂), 1.94–1.99 (m, 4H, 2CH₂), 2.97 (br, 1H, OH), 3.39–3.99 (br, 4H, 2CH₂), 4.00 (br, 1H, CH), 4.23–4.25 (br, 1H, CH), 4.85–4.89 (br, 1H, CH), 6.17 (d, ³J = 6.9, 1H, NH). ¹³C NMR (125.7 MHz, CDCl₃): δ 24.6 (2CH₂), 25.3 (2CH₂), 25.4 (2CH₂), 25.8 (CH₂), 25.9 (2CH₂), 29.5 (CH₂), 29.9 (CH₂), 30.3 (CH₂), 32.4 (CH₂), 32.5 (CH₂), 49.1 (CH), 56.2 (CH), 60.7 (CH), 178.2 (C=S), 184.4 (C=S).